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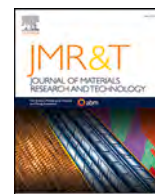
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Radiation response of multicomponent $L1_2 \gamma'$ precipitates strengthened high entropy alloys: The role of γ/γ' interface

Shasha Huang^a, Haijun Fu^a, Yaoxu Xiong^a, Wenyu Lu^a, Shihua Ma^a, Xuepeng Xiang^{a,b}, Ji-Jung Kai^a, Shijun Zhao^{a,*}

^a Department of Mechanical Engineering, City University of Hong Kong, Hong Kong, China

^b School of Environment and Energy, South China University of Technology, Guangzhou, 510006, China

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ABSTRACT

The advancement of high entropy alloys (HEAs) has stimulated the development of multicomponent $L1_2 \gamma'$ intermetallics, which exhibit both excellent strength and ductility. These intermetallics with long-range order are typically employed as coherent precipitates to enhance the high-temperature performance of HEAs. This study investigates the influence of multicomponent γ' intermetallics on the irradiation response of the strengthened HEA γ phase, focusing on the role of the multicomponent γ/γ' interface in defect production and evolution. Our results indicate that the chemical disorder within the multicomponent γ' phase leads to a broad defect energy spectrum near the interface zone, diminishing its effectiveness as a defect sink. Based on these observations, we propose that, distinct from traditional superalloys, the γ/γ' interface plays a minor role in the irradiation resilience of multicomponent γ' intermetallics-strengthened HEAs. Instead, the inherent chemical disorder within the multicomponent γ' intermetallics emerges as the key factor.

1. Introduction

The advancement in multiprincipal elemental alloys (MPEAs), including high-entropy alloys (HEAs), has marked a revolutionary approach in materials engineering, targeting exceptional mechanical properties and enhanced irradiation resistance [1–6]. In MPEAs, the inherent chemical disorder and resultant lattice distortion can effectively suppress defect kinetics, leading to high strength and promising radiation tolerance [7–12]. However, the application of HEAs faces the challenge of balancing strength and toughness at elevated temperatures [3,13–15]. A promising strategy to augment high-temperature performance involves the incorporation of the $L1_2 \gamma'$ intermetallics, known for their robust mechanical strength and prevalent use as strengthening components in conventional Ni-based superalloys [16–21]. Inspired by MPEAs, the introduction of multi-principal components into the sublattices of γ' intermetallics can significantly increase the anti-phase boundary energy, boosting their strengthening functionality [13]. By carefully selecting appropriate multicomponent intermetallics (MCIs) as precipitates, the properties of MCI-strengthened HEAs can be effectively tailored for high-temperature applications.

Ni-based superalloys find extensive utilization in nuclear reactors

due to their outstanding mechanical properties and corrosion resistance [22,23], rendering their irradiation response critical for reactor safety and performance. The beneficial properties of these alloys are primarily attributed to coherent $L1_2$ -structured γ' Ni_3Al precipitates dispersed throughout the γ -Ni matrix [24–26]. In such alloys, both the γ phase matrix and γ' intermetallics share a face-centered cubic (FCC) structure, fostering a coherent interface between the matrix and the precipitate, which is essential for strengthening purposes. Under irradiation, the γ/γ' interface, characterized by lattice mismatch due to differing lattice constants between γ and γ' phases, plays a vital role in defect dynamics. The γ' particles often undergo disorder and start dissolving at the interface under irradiation [27–31], impacting phase stability and the irradiation response of the γ' -strengthened superalloys [27,31]. The incorporation of multicomponents into the precipitates causes chemical disordering at the γ/γ' interface, potentially altering elemental arrangement and coherence. Consequently, the interface is anticipated to exert a significant impact on the irradiation response of MCI-strengthened HEAs.

Experiments on MCI-strengthened HEAs reveal a distinct role of γ/γ' interface under irradiation. For instance, MCI-strengthened NiFeCoCr HEA demonstrates a uniform distribution of helium bubbles across both

* Corresponding author.

E-mail address: shijzhao@cityu.edu.hk (S. Zhao).

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γ and γ' phases, suggesting a minor effect of the chemically disordered γ/γ' interface [32]. The chemical disorder in MCIs not only affects the degree of mismatch at interfaces but also alters the energy landscape for defect evolution. It has been revealed that partial chemical disorder within the γ' phase significantly influences defect diffusion properties [33,34]. Specifically, chemical disorder induced by Co atoms in $(\text{Ni}, \text{Co})_3\text{Al}$ has been observed to enhance vacancy diffusion, thus facilitating defect recombinations [33]. Therefore, in MCIs, the chemical disorder may override the role of the interface as defect sinks by modifying defect dynamics.

Building on prior research that underscores the impressive irradiation resistance of L1_2 -strengthened HEAs, such as $(\text{Ni}, \text{Co})_3(\text{Al}, \text{Ti})$ -strengthened NiFeCoCrAlTi [34], this study examines the role of the γ/γ' interface in MCI-strengthened HEAs under irradiation. We focus on the effects of interfacial mismatch and chemical disorder, considering several typical γ/γ' systems, including $\text{Ni}/\text{Ni}_3\text{Al}$, $\text{Ni}/\text{Ni}_3\text{Ti}$, $\text{Ni}/\text{Ni}_3(\text{Al}, \text{Ti})$, $\text{Ni}/(\text{Ni}, \text{Co})_3\text{Al}$, and $\text{Ni}/(\text{Ni}, \text{Co})_3(\text{Al}, \text{Ti})$, where Co and Ti are critical elements assisting the precipitation of γ' phase [31,35,36]. Although Co is readily activated by thermal neutrons, it has a low fast neutron absorption cross-section [37], making it viable for use in fast neutron reactors. Additionally, non-equilibrium conditions such as irradiation-enhanced diffusion and irradiation-induced segregations [38,39] significantly promote local chemical ordering within HEAs, which can lead to defect trapping in localized areas, such as pure Ni [40]. In this context, our study models the γ matrix exclusively as pure Ni to streamline the analysis of interfacial mismatch effects. Through molecular dynamics (MD) simulations, we demonstrate that while increasing interfacial mismatch can enhance the ability of the interface to act as a defect sink, the presence of chemical disorder can dilute this effect by generating a rough defect energy landscape near the interface, diminishing the role of interfaces as defect recombination catalysts. This study highlights a strategic balance of interfacial mismatch and chemical disorder for optimizing the irradiation resistance of nano-precipitate strengthened HEAs.

2. Method

MD and molecular statics (MS) simulations were performed using LAMMPS [41]. The Modified Embedded Atom Model (MEAM) developed in our previous work was employed [42]. MD simulations were conducted in a $34a_0 \times 34a_0 \times 34a_0$ FCC supercell [43–45], while a $10a_0 \times 10a_0 \times 20a_0$ FCC supercell was used for MS, where a_0 denotes the lattice constant of the system. In both cases, the supercells were structured such that the upper half contained the γ' phase and the lower half consisted of pure Ni, with both maintaining coherent $\{100\}$ crystallographic planes to form the $[100]_{\gamma'}/[100]_{\text{Ni}}$ interface. Partial chemical disorder in the L1_2 A_3B intermetallic was introduced by randomly distributing Ni and Co atoms within the A sublattice, and Al and Ti within the B sublattice, following the stoichiometries of $\text{Ni}_3(\text{Al}, \text{Ti})$, $(\text{Ni}, \text{Co})_3\text{Al}$, and $(\text{Ni}, \text{Co})_3(\text{Al}, \text{Ti})$, respectively. For cascade simulations, a primary knock-on atom (PKA) with an energy of 5 keV was introduced at various initial distances from the interface and directed towards it to simulate an ion bombardment event. The variable timestep method was applied, dynamically adjusting the timestep to restrict atomic displacements to a maximum of 0.01 Å per step. The total simulation time was approximately 50 ps to allow for the complete cooling of the damage cascade. To prevent boundary effects, the border region (two atomic layers around the simulation cell) was maintained at 300 K through a Nose-Hoover thermostat. The resulting defects were examined using the Wigner-Seitz defect analysis as implemented in the Ovito software [46].

The defects considered in this work include interstitials and vacancies, with their formation energies calculated using $E_f = E - E_0 \pm \mu$, where E and E_0 are the total energies of the supercell with and without defects, respectively, and μ is the chemical potential of the element

composed of the defect (energy per atom in the corresponding pure metals). The vacancy migration energy was determined by the climbing-image nudged elastic band (CI-NEB) method [47], with a timestep of 0.01 fs, and a total of 15 intermediate images. The force convergence criterion in NEB calculations was 1×10^{-6} eV/Å. The interfacial mismatch is quantified by $\epsilon = 2 \frac{a_+ - a_-}{a_+ + a_-}$ [48], where a_+ and a_- represent the crystal lattice parameters of the upper (γ') and lower (γ) phases, respectively.

To compare the diffusion properties of interstitials and vacancies across different systems, the trajectory of a single Ni interstitial or a single vacancy within 30 ns is analyzed. Considering the lower mobility of vacancy compared to interstitial, the diffusion of vacancy was examined at 1400 K, whereas that of interstitial was at 1000 K.

3. Results

3.1. Interfacial mismatch

The calculated interfacial mismatches for each considered system are displayed in Table 1. Notably, the interfacial mismatch of the $\text{Ni}/\text{Ni}_3\text{Al}$ system corroborates well with previous first-principles calculations and experimental measurements [49–52]. Despite their relatively simple chemical compositions, $\text{Ni}/\text{Ni}_3\text{Al}$ and $\text{Ni}/\text{Ni}_3\text{Ti}$ show significantly distinct interfacial mismatches, providing a foundation to explore their impact on defect behaviors. Meanwhile, $\text{Ni}/\text{Ni}_3(\text{Al}, \text{Ti})$ and $\text{Ni}/(\text{Ni}, \text{Co})_3\text{Al}$ have intermediate levels of mismatch, despite their increased chemical complexity. In contrast, $\text{Ni}/(\text{Ni}, \text{Co})_3(\text{Al}, \text{Ti})$ exhibits the highest degrees of both chemical complexity and interfacial mismatch among the systems examined.

3.2. Primary damage production and recombination

In our cascade simulations, the PKA was initiated from the Ni side, as illustrated in Fig. 1(a). After a relaxation period of 50 ps, the number of survived defects in both the Ni and γ' sides was computed. Assuming the defect numbers are n_{Ni} and $n_{\gamma'}$ in the Ni and γ' phase, respectively, the fraction of defects produced in the mixed system relative to the defects in their respective pure phases was calculated as follows [53]:

$$f_{\text{defect}} = \frac{n_{\text{Ni}}}{n_{\text{Ni}}^{\text{bulk}}} + \frac{n_{\gamma'}}{n_{\gamma'}^{\text{bulk}}} \quad (2)$$

where $n_{\text{Ni}}^{\text{bulk}}$ and $n_{\gamma'}^{\text{bulk}}$ are the defect numbers produced in the pure Ni and γ' bulk phases, respectively. The average defect numbers from 200 independent simulations in the pure Ni and the γ' phases are listed in Table 2. Generally, more defects are observed in the ordered γ' phase than in pure Ni, likely due to the suppressed defect recombination, consistent with prior research [54].

Based on these numbers, the defect fractions for vacancies (VAC) and interstitials (SIA) as a function of the initial PKA position relative to the interface are depicted in Fig. 1(b)–(f). To ensure accuracy and account for the statistical error, 15 independent cascades were conducted at each PKA distance. The averaged defect fractions offer insights into the defect distribution relative to the pure phases and elucidates the role of the interface. Fig. 1(b)–(f) show that when the PKA is initiated close to the interface, the fractions of vacancies and interstitials are nearly equal, indicating that the coherent interface does not preferentially retain one defect type over the other.

Table 1
Interfacial mismatch (ϵ : %) between Ni and γ' phases.

	Ni/ Ni_3Al	Ni/ Ni_3Ti	Ni/ $\text{Ni}_3(\text{Al},$ Ti)	Ni/ $(\text{Ni},$ $\text{Co})_3\text{Al}$	Ni/ $(\text{Ni}, \text{Co})_3(\text{Al},$ Ti)
(ϵ : %)	1.37	2.10	1.81	2.08	2.14

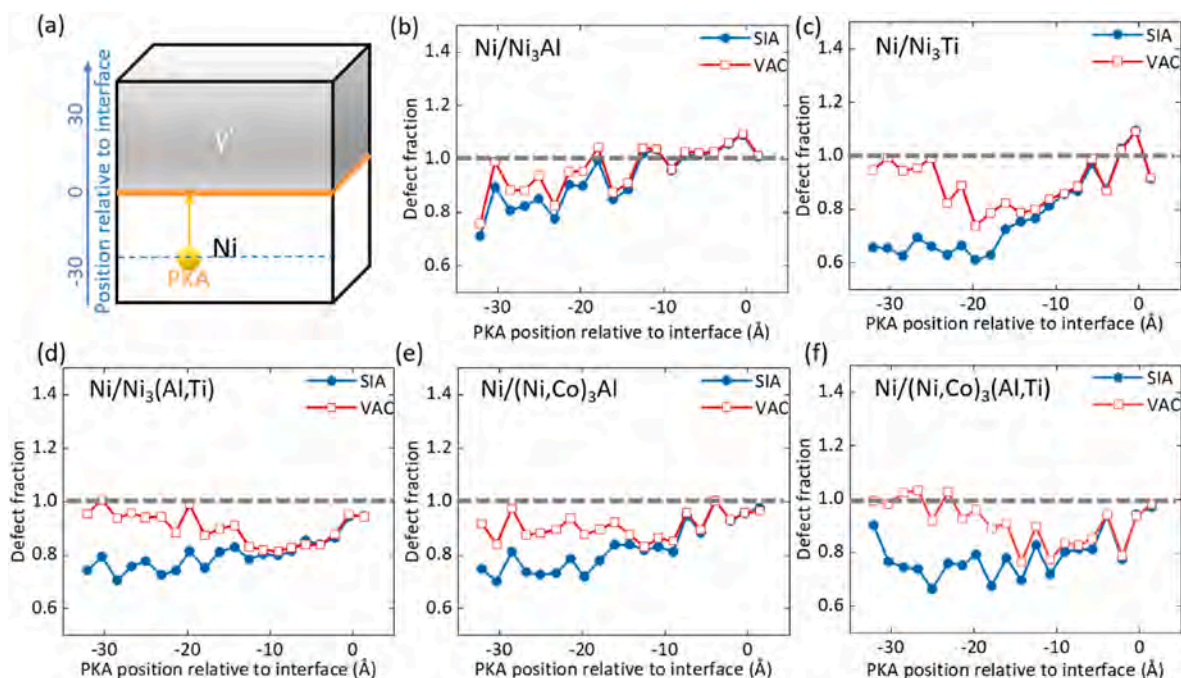


Fig. 1. (a) Schematic illustration of the cascade simulation model. The shadow region represents the γ' phase, and the yellow ball represents the Ni PKA initiated at different original positions with respect to the interface from the Ni side. The direction of the PKA is toward the interface from Ni to the γ' phase. Defect fraction produced in (b) Ni/Ni₃Al, (c) Ni/Ni₃Ti, (d) Ni/Ni₃(Al,Ti), (e) Ni/(Ni,Co)₃Al, and (f) Ni/(Ni,Co)₃(Al,Ti) as a function of PKA position relative to the interface. (For interpretation of the references to color in this figure legend, the reader is referred to the Web version of this article.)

Table 2

The average number of defects produced (n^{bulk}) in pure Ni and γ' phase derived from 200 independent 5 keV cascade simulations at 300 K.

	Ni	Ni ₃ Al	Ni ₃ Ti	Ni ₃ (Al,Ti)	(Ni,Co) ₃ Al	(Ni,Co) ₃ (Al,Ti)
n^{bulk}	16	20	41	28	28	31

However, when the PKA initiates further from the interface, a distinct difference emerges, with a higher fraction of vacancies compared to interstitials. This disparity arises from the greater mobility of interstitials compared to vacancies at the simulation temperature of 300 K. While vacancies remain predominantly localized near their production sites in the Ni region, the more mobile interstitials are capable of traversing into the γ' phase. Quantitative analysis, as shown in Table 2, demonstrates that the γ' phase consistently exhibits a higher average defect concentration than pure Ni. According to Eq. (2), the migration of interstitials into the γ' phase results in a reduced defect count of $n_{\gamma'}$ compared to defects generated directly within the γ' phase. This movement of interstitials between the two phases leads to a reduction in f_{defect} . Conversely, vacancies are more likely to stay near their formation locations, displaying fractions close to unity except in areas where interface effects are pronounced.

The defect fraction value in the mixed system serves as an indicator of defect recombination. Particularly, $f_{defect} < 1$ suggests that the interface enhances defect recombination compared to pure phases, while a value close to 1 implies the minimal impact of the interface on recombination. The analysis of Fig. 1(b) and (c) reveals that a higher interfacial mismatch in Ni/Ni₃Ti enhances defect recombination, especially when the cascade core region is close to the interface (with PKA position about 15 Å from the interface). In contrast, the Ni/Ni₃Al interface exhibits a much weaker influence, as evidenced by the near-one f_{defect} values consistently.

The interfacial mismatch in Ni/Ni₃(Al,Ti) and Ni/(Ni,Co)₃Al is stronger than that in Ni/Ni₃Al, leading to more efficient defect recombination, as illustrated in Fig. 1(d) and (e). Despite Ni/(Ni,Co)₃(Al,Ti)

having the most severe interfacial mismatch among the systems studied, its defect behavior is similar to that of Ni/Ni₃(Al,Ti) and Ni/(Ni,Co)₃Al. Compared to Ni/Ni₃Ti, Ni/(Ni,Co)₃(Al,Ti) demonstrates a weaker interface effect on defect annealing, suggesting that the incorporation of chemical disorder may diminish the impact of interfacial mismatch on defect evolution. Nevertheless, for all the MCIs, f_{defect} values remain below one, indicating that the interfaces, despite their reduced effect, still help to decrease defect numbers relative to pure phases. This aligns with observations from previous studies showing defect suppression in HEAs with γ' phase [32,55,56].

3.3. Defect energetics

To further explore the interplay between interfacial mismatch and chemical disorder observed in Fig. 1, we compare the defect formation energies in Ni/Ni₃Al, Ni/Ni₃Ti, and Ni/(Ni,Co)₃(Al,Ti) by manually introducing a single $\langle 100 \rangle$ dumbbell or vacancy into the $10a_0 \times 10a_0 \times 20a_0$ FCC supercell. Here, Ni/Ni₃Al and Ni/Ni₃Ti were selected for their differing interfacial mismatches but similar chemical disorder, providing a contrast in structural but not compositional complexity. Conversely, Ni/(Ni,Co)₃(Al,Ti) was chosen for its interfacial mismatch similar to that of Ni/Ni₃Ti, yet featuring distinct chemical disorder, allowing us to isolate and evaluate the effects of chemical complexity on defect formation energies.

Fig. 2(a)–(c) provide the formation energies of $\langle 100 \rangle$ dumbbell, while Fig. 2(d)–(f) present the vacancy formation energies relative to the interface position. Given the chemical complexity of Ni/(Ni,Co)₃(Al,Ti), 100 independent calculations were conducted to obtain the average E_f , along with error bars indicating the standard error in Fig. 2(c) and (f). In L1₂ unit cells, two inequivalent Ni sites at the same (001) plane result in distinct E_f for Ni-containing dumbbells. The results of interstitial formation energies reveal lower values in the γ' phase, aligning with the preference for interstitials in this phase, as discussed in Fig. 1.

In Ni/Ni₃Al, the vacancy formation energies remain consistent across the interface, as demonstrated in Fig. 2(d). Notably, Fig. 2(e) indicates a significant decrease in vacancy formation energies from the Ni side to

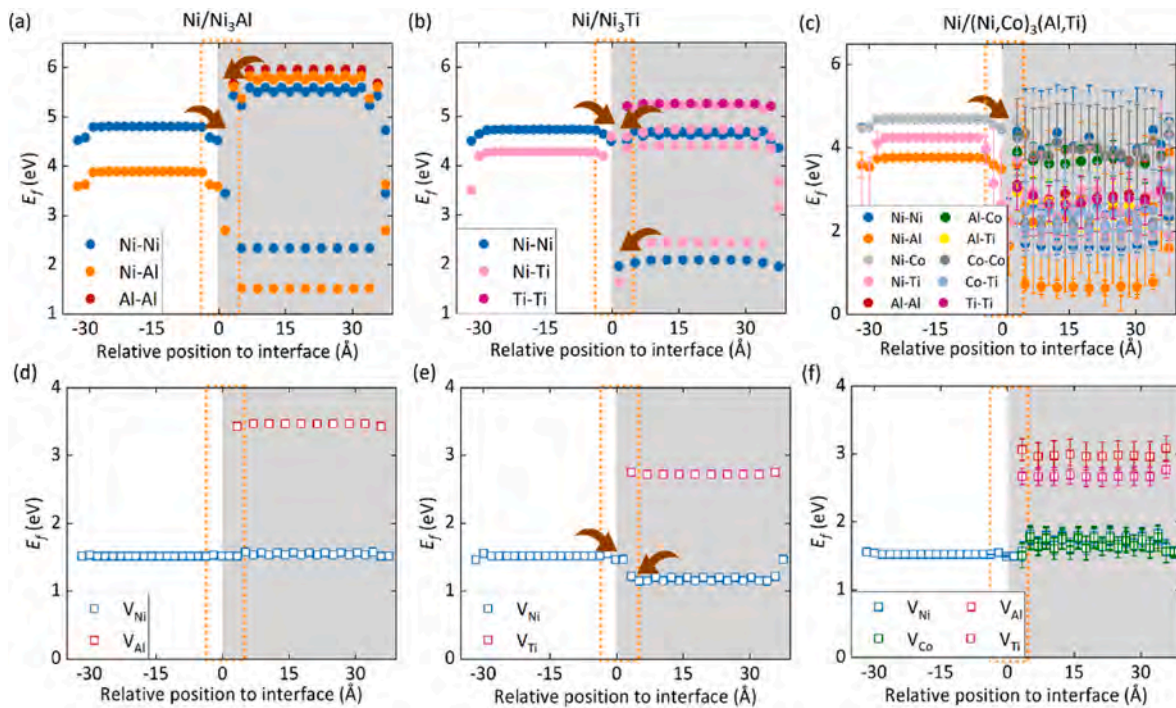


Fig. 2. Formation energies of a $\langle 100 \rangle$ dumbbell in (a) Ni/Ni₃Al, (b) Ni/Ni₃Ti, and (c) Ni/(Ni,Co)₃(Al,Ti); and of a vacancy in (d) Ni/Ni₃Al and (e) Ni/Ni₃Ti, and (f) Ni/(Ni,Co)₃(Al,Ti) as a function of the relative position to the interface. The shadow regions represent the γ' phase, and the brown arrows highlight the trends in formation energy near the interface. (For interpretation of the references to color in this figure legend, the reader is referred to the Web version of this article.)

the γ' side in Ni/Ni₃Ti, with both sides reaching minimum values at the interface (highlighted by the brown arrows around the orange dashed box). This trend is mirrored for interstitials, suggesting that the severe interfacial mismatch improves the role of the Ni/Ni₃Ti interface as a defect sink, enhancing defect annihilation, as observed in Fig. 1(b).

Despite a similar interfacial mismatch to Ni/Ni₃Ti, the presence of chemical disorder in Ni/(Ni,Co)₃(Al,Ti) introduces a diverse local chemical environments, resulting in a broader range of defect formation energies in both γ' side and interface region, as illustrated by the energy distributions in Fig. 1(c) and (f). Consequently, unlike in Ni/Ni₃Ti, where vacancies and interstitials encounter consistently low-energy sites at the interface, the broad energy distribution around the interface region of Ni/(Ni,Co)₃(Al,Ti) decreases the likelihood of defects being captured and annihilated. This diminished defect-sink effectiveness highlights the impact of chemical disorder on defect behavior at the interface.

3.4. Defect dynamics

In addition to defect formation energies, defect dynamics play a critical role in determining how defects interact with interfaces [54,57].

To capture these dynamics, we calculated vacancy migration barriers in different zones, including the γ' side, the interface zone, and the Ni side. Owing to the crystalline symmetry in Ni/Ni₃Al and Ni/Ni₃Ti, there are three different migration paths in γ' side and interface zone, respectively. In contrast, 25 independent NEB calculations were performed in Ni/(Ni,Co)₃(Al,Ti) in each zone (Fig. 3(c)) to consider the chemical complexity. The average migration energies in different zones are also indicated in Fig. 3.

In the Ni/Ni₃Al system, the vacancy migration energy increases from the Ni side to the interface, followed by a slight decrease on the γ' side. However, in Ni/Ni₃Ti, the vacancy migration energy is significantly reduced at the interface zone, suggesting enhanced defect transport through the Ni/Ni₃Ti interface. The Ni/(Ni,Co)₃(Al,Ti) system, characterized by higher chemical complexity, displays similar migration energy distributions in the interface zone and γ' phase, indicating a diminished interfacial effect.

Our cascade simulation results in Fig. 1 indicate that interstitials tend to migrate towards the γ' phase. To further understand the mechanism, we directly simulated the diffusion behavior of a single Ni interstitial or vacancy in the three interface-containing systems. Given that vacancies diffuse slowly at low temperatures, the diffusion temperature was set to

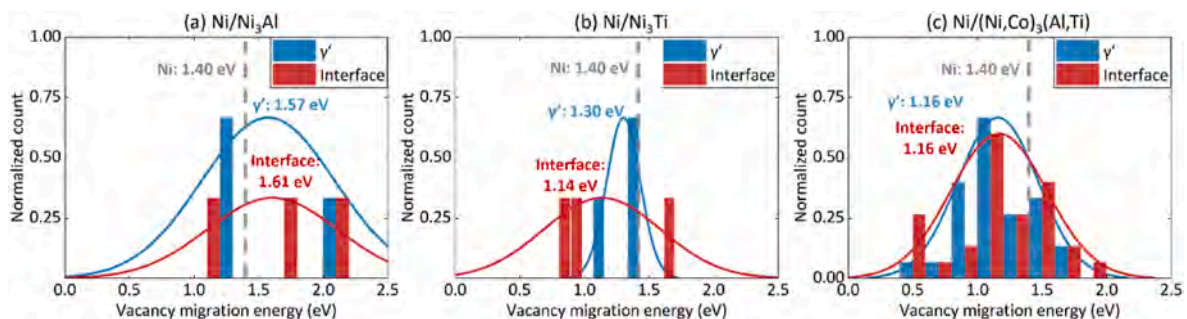


Fig. 3. Vacancy migration energies in (a) Ni/Ni₃Al, (b) Ni/Ni₃Ti, and (c) Ni/(Ni,Co)₃(Al,Ti). The average migration energies in different zones were also shown in the figure.

1400 K for vacancies and 1000 K for interstitials to accelerate the simulation process. Fig. 4 depicts the trajectories of the z-coordinate (the normal direction to the interface) of the defects.

In these simulations, defects were initialized at three distinct locations: (i) in the γ' phase (red dots), (ii) near the interface (green dots), and (iii) in the Ni phase (blue dots). To account for statistical variability, three independent simulations were conducted for vacancies at each initial position (represented by different shades of color). In the Ni/Ni₃Al system, vacancies slightly favor the Ni phase, while all interstitials gravitate towards the Ni₃Al side. The divergence in their trajectories diminishes the probability of mutual interaction with the interface, resulting in a defect fraction close to one for all cases examined (Fig. 1(b)).

In contrast, both vacancies and interstitials in the Ni/Ni₃Ti system show a preference for residing within the Ni₃Ti phase, irrespective of their initial positions. The concurrent diffusion of both vacancies and interstitials toward the γ' phase increases their likelihood of mutual interaction with the interface. This behavior facilitates defect recombination around the interface, consistent with the lower than one defect fraction value observed in the Ni/Ni₃Ti system (as shown in Fig. 1(c)).

In the Ni/(Ni,Co)₃(Al,Ti) system, the reduced difference in vacancy migration barriers between the interface and the γ' phase diminished the role of interface in vacancy migration. Furthermore, the rough energy landscape induces local trapping of vacancies, causing them to oscillate near the interface. Examination of Fig. 4(d)–(f) reveals that interstitials in Ni/(Ni,Co)₃(Al,Ti) are confined to a more restricted area away from the interface than Ni/Ni₃Al and Ni/Ni₃Ti, possibly due to the broad distribution of interstitial formation energies, with interstitials preferentially occupying locations of lower formation energy. The attenuated tendency for vacancy transport into the γ' phase, coupled with reduced interstitial-interface interactions, resulted in a slightly diminished capacity of the interface to promote defect recombination than Ni/Ni₃Ti.

Although the diffusion trajectories were simulated at only one temperature for each defect type, the results in Fig. 4 indicate that their preferences for different regions at high temperatures are strongly influenced by their formation energies and migration energy barriers at 0 K. This suggests that the energy landscape, rather than temperature,

plays a more dominant role in determining their behavior, indicating that the observed diffusion patterns may be broadly applicable at other temperatures.

The above findings confirm that in both Ni/Ni₃Al and Ni/Ni₃Ti, interstitials generated by irradiation favor the γ' side, while the behavior of vacancies is influenced by the energy landscape. The pronounced interfacial mismatch can enhance the sink effect, promoting more efficient vacancy migration and potentially enhancing defect recombination at the interface. However, in Ni/(Ni,Co)₃(Al,Ti), chemical disorder reduces the transport of vacancies into the γ' phase, while simultaneously reducing interstitial-interface interactions. This trend implies that chemical disorder may diminish the impact of interfacial mismatch, potentially reducing the effectiveness of the interface in aiding defect migration and recombination.

3.5. Defect evolution at kinetic stage

The prior single cascade simulation allows for the simultaneous observation of how interfaces affect defect generation and recombination, primarily resulting from interstitial diffusion. To further explore how interfacial mismatch and chemical disorder influence defect evolution involving vacancies, we introduced Frenkel pairs (FPs) by randomly displacing atoms within the cells at a defect concentration of 0.5%. Their clustering and diffusion behaviors were then monitored over 1 ns at 1000 K using the NVT ensemble.

Fig. 5 showcases the defect configurations at the end of the simulations. Despite the initially uniform distribution of defects, interstitials rapidly concentrated in the γ' side, expediting the recombination of FPs and reducing the number of vacancies in γ' side, leading to the separated distribution of interstitials and vacancies. However, the mismatched interfaces significantly enhanced vacancy transport and facilitated recombination, notably decreasing the defect density near the Ni/Ni₃Ti interface and forming a nearly denuded zone, as marked by grey dashed box in Fig. 5(b). This effect was less evident in Ni/Ni₃Al due to its weaker mismatch. In the more chemically complex Ni/(Ni,Co)₃(Al,Ti), despite a higher mismatch than Ni/Ni₃Ti, a broader energy spectrum reduced the effectiveness of interface as a defect sink, leading to a diminished

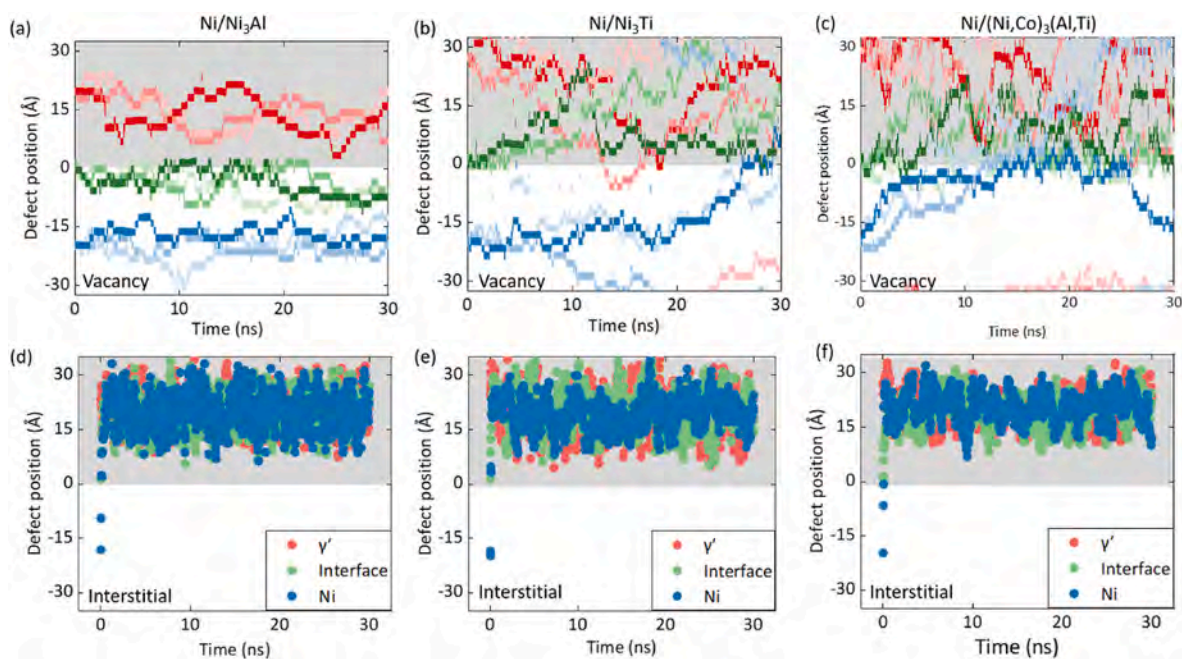


Fig. 4. Trajectories of the z coordinate (normal direction to the interface) for a vacancy in (a) Ni/Ni₃Al, (b) Ni/Ni₃Ti, and (c) Ni/(Ni,Co)₃(Al,Ti); and for an interstitial in (d) Ni/Ni₃Al, (e) Ni/Ni₃Ti, and (f) Ni/(Ni,Co)₃(Al,Ti). The shadow region represents the γ' phase. Vacancy diffusion is simulated at 1400 K, whereas that of interstitial is at 1000 K.

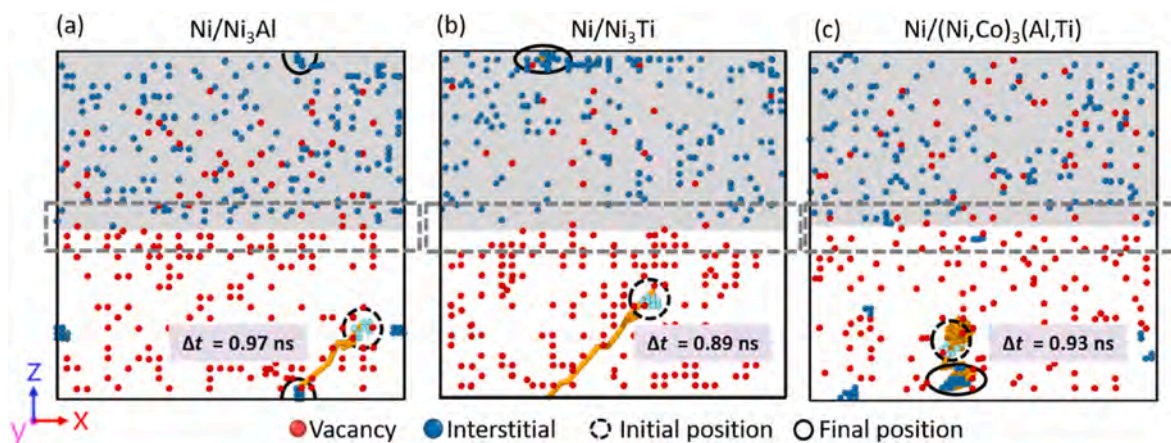


Fig. 5. Defect configuration and trajectory of interstitial clusters in (a) Ni/Ni₃Al, (b) Ni/Ni₃Ti, and (c) Ni/(Ni,Co)₃(Al,Ti) at the end of the simulation ($t = 1$ ns). The shadow region represents the γ phase. The initial position of the interstitial cluster when it was formed is encircled by black dashed circles, while its final position is shown by solid circles. The trajectory of the cluster is illustrated by orange lines. The duration of each trajectory is also indicated in the figure. (For interpretation of the references to color in this figure legend, the reader is referred to the Web version of this article.)

interface effect than Ni/Ni₃Ti.

The weakened interface effect by chemical complexity can also be observed from the diffusion trajectories of interstitial clusters. In all three cases, interstitial clusters migrated towards the interface, consistent with previous experimental observations of interstitial clusters formation at interfaces [29,58]. Fig. 5(a) and (b) illustrate a more pronounced attractive effect at the highly mismatched Ni/Ni₃Ti interface than the Ni/Ni₃Al interface, whereas the trajectories in Ni/(Ni,Co)₃(Al, Ti) were more erratic, suggesting that chemical disorder suppresses the defect-interface interaction. Notably, the migration duration of the cluster towards the Ni/Ni₃Ti interfaces is the shortest, as indicated by the values of 0.89 ns. This fast migration suggests strong defect-interface coupling in this system. In contrast, the longer duration of 0.93 ns in Ni/(Ni,Co)₃(Al,Ti) reflects more diffusive and less directed migration behavior, further elucidating that chemical disorder weakens the interaction between defects and the interface.

4. Discussion

In reality, the incorporation of L1₂ γ' intermetallics effectively enhances the high-temperature strength of HEAs [16–19], but the inherent chemical complexity of HEAs often leads to chemical disorder within these intermetallics. This disorder, along with lattice mismatches

between the γ' phase and the matrix, significantly affects how interfaces interact with defects, thereby influencing the irradiation response of precipitate-strengthened HEAs [59–61]. Our analysis of traditional γ' phases (Ni₃Al and Ni₃Ti) shows that an increased interfacial mismatch promotes vacancy migration and interstitial absorption at the interface, enhancing FPs recombination, at least at the initial stage of irradiation damage when irradiation-induced ballistic mixing is dominant. These findings aligned with previous studies show that Ni₃Ti-strengthened superalloys exhibit a higher threshold dose for disordering compared to those strengthened with Ni₃Al at similar temperatures [30,42,62,63]. Nonetheless, the appearance of chemical disorder broadens the defect energy spectrum around the interface, thereby diminishing its effectiveness in annihilating defects. A schematic illustration of these processes is provided in Fig. 6. Although the HEA matrix in this study is simplified as Ni, the presence of multiple principal elements in HEAs generally creates a complex energy landscape that can trap defects in local regions. Consequently, similar weakening effects of interfaces are expected, consistent with the uniform distribution of defects found in nanoparticle-strengthened HEAs [32,34,64,65], thus suggesting a broader applicability of our findings across diverse HEA systems.

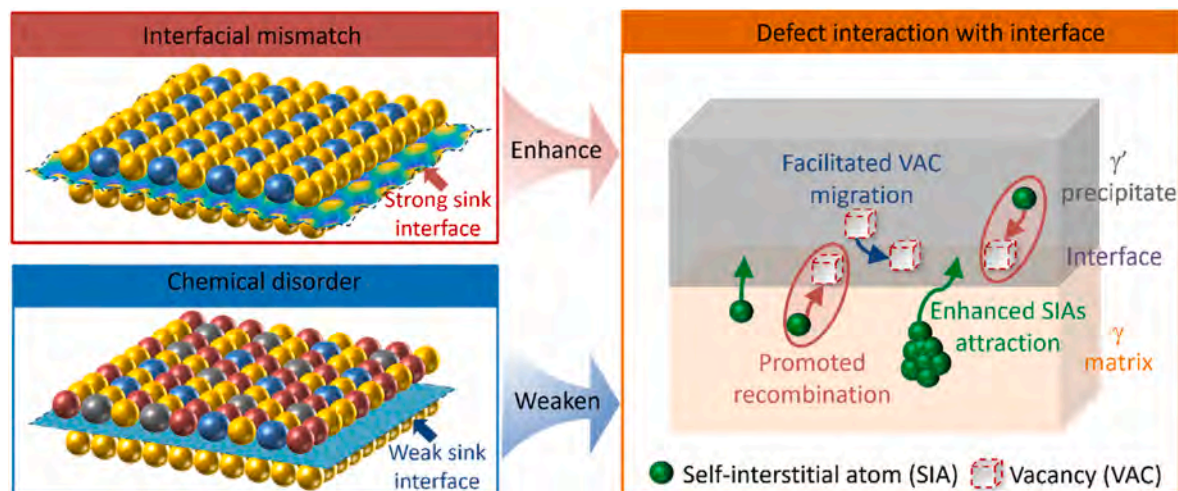


Fig. 6. Schematic illustration of the influence of interfacial mismatch and chemical disorder on defect evolution in γ/γ' system.

5. Conclusion

In summary, we investigate the irradiation response of multicomponent intermetallics-strengthened HEAs, focusing on the role of the γ/γ' interface. Through atomistic simulations, we assess how interfacial mismatch and chemical disorder impact defect production and evolution. Our results show that a highly mismatched interface acts as an effective defect sink, enhancing defect recombination. However, the introduction of chemical disorder reduces the effectiveness of interfaces by creating a broadly distributed energy spectrum, even in the presence of significant mismatch. This study thus highlights the complex interplay between interfacial mismatch and chemical disorder in MCI-strengthened HEAs, revealing their nuanced impact on material resilience under irradiation.

Declaration of competing interest

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

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